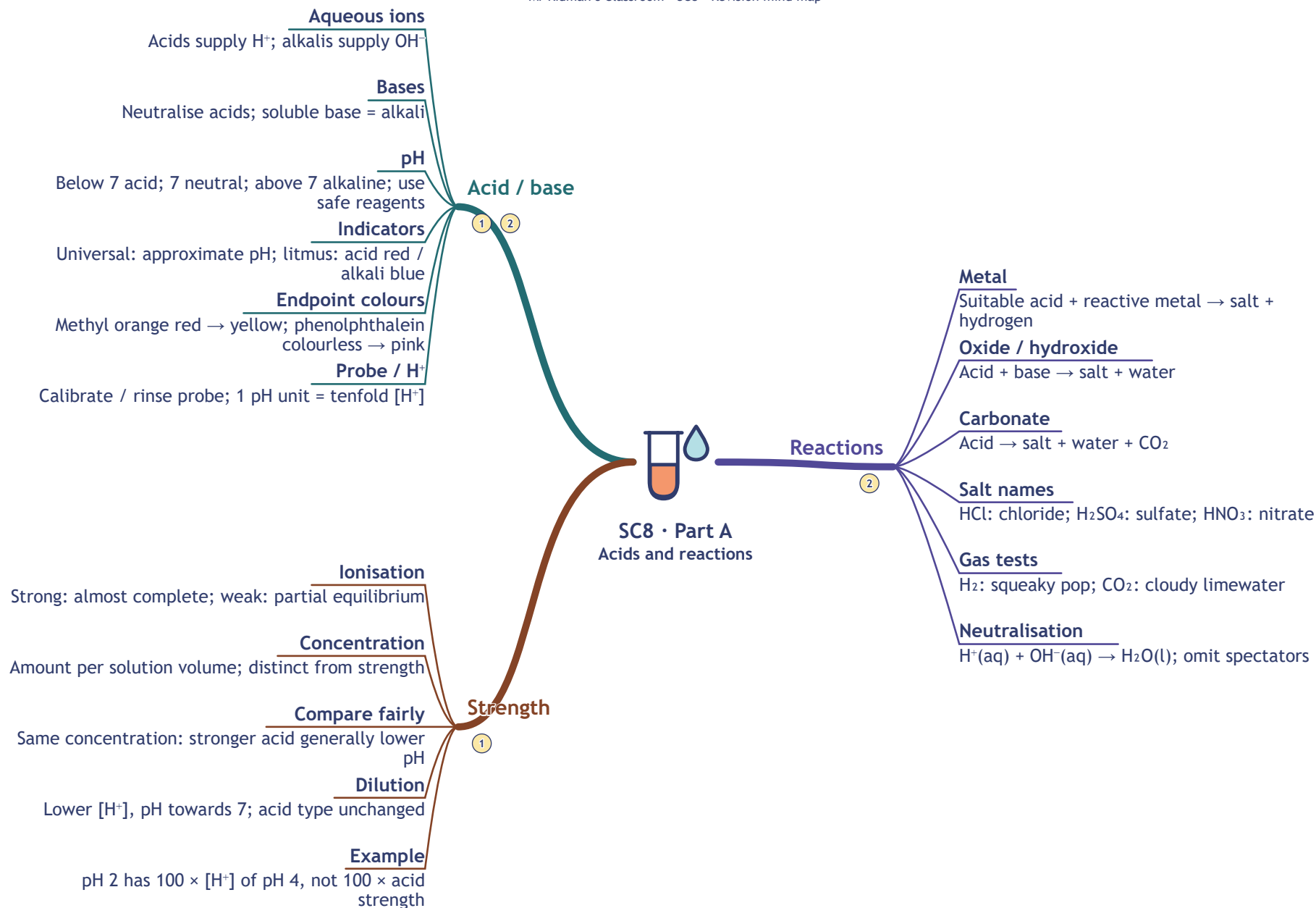




1. Acid / base → Strength: pH depends on hydrogen-ion concentration.
2. Acid / base → Reactions: Neutralisation removes H^+ with a base.